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## Key indicators

Single-crystal X-ray study  
 $T = 273$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.066  
 $wR$  factor = 0.148  
Data-to-parameter ratio = 12.2For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## 1-(1,8-Dihydroxy-6-methoxy-3-methyl-naphthalen-2-yl)ethanone

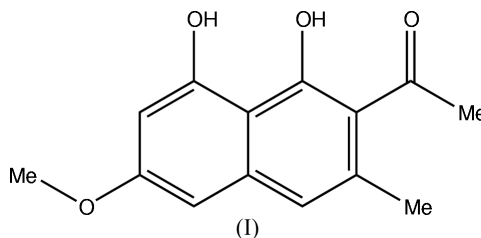
The molecule of the title compound,  $\text{C}_{14}\text{H}_{14}\text{O}_4$ , is planar and the structural dimensions are in the normal ranges. There are two intramolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds, but no intermolecular interactions were observed.

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## Comment

Several naphthalene derivatives, such as torachryson and torachryson-8-*O*- $\beta$ -*D*-glucopyranoside, have been isolated from Rumex (Demirezer *et al.*, 2001; Suri *et al.*, 1978) and Rhubarb (Tsuboi *et al.*, 1977) plant species in Turkey and Japan, respectively. The title compound, (I), extracted from the *Cassia alata* plant found in West Malaysia, is an analogue of torachryson but with a methyl group at C1 on the naphthalene fragment (Fig. 1). Genus *Cassia* is known to possess some medicinal properties and is a rich source of anthraquinones and flavonoids (Gupta & Singh, 1991). Therefore, *Cassia alata* is likely to be only the second species, in addition to *C. tora* (Shibata *et al.*, 1969), found to contain compound (I).The molecule is planar apart from methyl H atoms, with a maximum deviation from the mean plane of  $-0.025$  (4) Å for atom C14 from the mean plane. The bond lengths and angles (Table 1) are in the normal ranges (Allen *et al.*, 1987) and are comparable to those in chemically synthesized 2-bromo-1-(1-hydroxynaphthalen-2-yl)ethanone (Köysal *et al.*, 2004). However, in (I), the C9–O3 bond [1.335 (3) Å] is slightly shorter than C7–O2 [1.356 (3) Å], which in turn is comparable to the length of 1.366 (7) Å observed in 2-bromo-1-(1-hydroxynaphthalen-2-yl)ethanone. There are two intramolecular hydrogen bonds, *viz.* O2–H2A $\cdots$ O3 and O3–H3A $\cdots$ O4 (Table 2), resulting in the formation of two pseudo-six-membered rings (C9–C8–C7–O2–H2A $\cdots$ O3 and C13–C10–C9–O3 $\cdots$ H). In contrast to 2-bromo-1-(1-hydroxynaphthalen-2-yl)ethanone, there are no intermolecular interactions observed in this structure.

## Experimental

*Cassia alata* was collected from Section 16 Shah Alam, Selangor, in September 2002. The specimen (SK996-04) was deposited at the

Laboratory of Natural Products Mini Herbarium, Institute of Bioscience, Universiti Putra Malaysia. The bark of *C. alata* (1.0 kg) was extracted with dichloromethane at room temperature for 2 d. Removal of solvent gave 12 g of crude extract which was fractionated on a silica gel column, eluting with dichloromethane containing increasing percentages of acetone. Fractions 4 to 9, which eluted at 90:10 dichloromethane–acetone, were rechromatographed to give 25 mg of (I) (long yellow needles; m.p. 483–487 K) after evaporation. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz): δ 2.64 (3H, s, CH<sub>3</sub>), 2.75 (3H, s, CH<sub>3</sub>), 3.90 (3H, s, OCH<sub>3</sub>), 6.85 (1H, s), 6.51 (1H, dd, *J*<sub>m</sub> = 2.4 Hz), 6.48 (1H, dd, *J*<sub>m</sub> = 2.4 Hz), 10.44 (1H, s, OH); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz): δ 203.6 (C=O), 169.7 (C1), 134.3 (C2), 139.9 (C3), 121.0 (C4), 99.2 (C5), 163.6 (C6), 100.8 (C7), 160.2 (C8), 108.8 (C9), 112.3 (C10), 25.4 (CH<sub>3</sub>), 31.8 (CH<sub>3</sub>) and 55.4 (OCH<sub>3</sub>). EI-MS *m/z*: 246 (*M*<sup>+</sup>), 231 (100).

Crystal data

C <sub>14</sub> H <sub>14</sub> O <sub>4</sub>	Z = 2
<i>M</i> <sub>r</sub> = 246.25	<i>D</i> <sub>x</sub> = 1.441 Mg m <sup>-3</sup>
Triclinic, <i>P</i> $\bar{1}$	Mo <i>K</i> α radiation
<i>a</i> = 6.993 (4) Å	Cell parameters from 518 reflections
<i>b</i> = 7.543 (4) Å	<i>θ</i> = 1.8–25.0°
<i>c</i> = 11.058 (6) Å	<i>μ</i> = 0.11 mm <sup>-1</sup>
<i>α</i> = 83.530 (9)°	<i>T</i> = 273 (2) K
<i>β</i> = 85.171 (9)°	Slab, yellow
<i>γ</i> = 79.017 (9)°	0.50 × 0.18 × 0.16 mm
<i>V</i> = 567.7 (5) Å <sup>3</sup>	

Data collection

Bruker SMART APEX area-detector diffractometer	1981 independent reflections
<i>ω</i> scans	1368 reflections with <i>I</i> > 2σ( <i>I</i> )
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	<i>R</i> <sub>int</sub> = 0.030
<i>T</i> <sub>min</sub> = 0.94, <i>T</i> <sub>max</sub> = 0.98	<i>θ</i> <sub>max</sub> = 25.0°
5437 measured reflections	<i>h</i> = -8 → 8
	<i>k</i> = -8 → 8
	<i>l</i> = -13 → 13

Refinement

Refinement on <i>F</i> <sup>2</sup>	<i>w</i> = 1/[σ <sup>2</sup> ( <i>F</i> <sub>o</sub> <sup>2</sup> ) + (0.05 <i>P</i> ) <sup>2</sup> + 0.2434 <i>P</i> ]
<i>R</i> [ <i>F</i> <sup>2</sup> > 2σ( <i>F</i> <sup>2</sup> )] = 0.066	where <i>P</i> = ( <i>F</i> <sub>o</sub> <sup>2</sup> + 2 <i>F</i> <sub>c</sub> <sup>2</sup> )/3
<i>wR</i> ( <i>F</i> <sup>2</sup> ) = 0.148	(Δ/ <i>σ</i> ) <sub>max</sub> < 0.001
<i>S</i> = 1.14	Δ <i>ρ</i> <sub>max</sub> = 0.23 e Å <sup>-3</sup>
1981 reflections	Δ <i>ρ</i> <sub>min</sub> = -0.14 e Å <sup>-3</sup>
163 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

O1—C5	1.365 (3)	O3—C9	1.335 (3)
O1—C12	1.428 (3)	O4—C13	1.259 (3)
O2—C7	1.356 (3)		
C12—O1—C5—C4	-1.7 (4)	C9—C10—C13—O4	0.8 (4)
C12—O1—C5—C6	177.9 (3)	C1—C10—C13—O4	-178.5 (3)

Table 2

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O2—H2 <i>A</i> ...O3	0.82	1.85	2.570 (3)	145
O3—H3 <i>A</i> ...O4	0.82	1.65	2.400 (3)	150

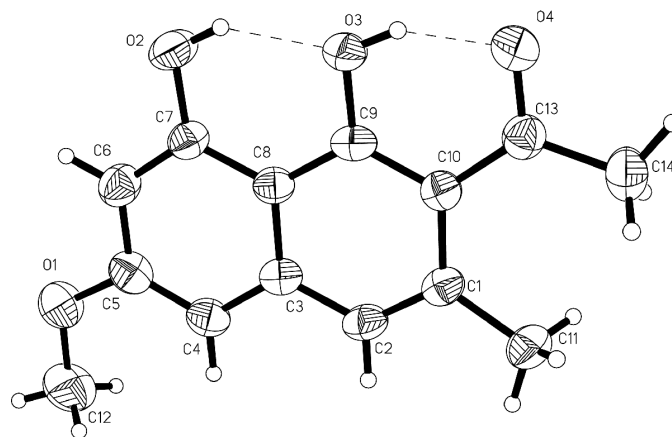


Figure 1

The molecular structure of the title compound, (I), shown with 50% probability displacement ellipsoids and H atoms as spheres of arbitrary radii. The dashed lines indicate hydrogen bonds.

H atoms were located in a difference map, idealized and refined as riding atoms, with C—H distances of 0.93 Å, methyl C—H distances of 0.96 Å and O—H distances of 0.82 Å, and with *U*<sub>iso</sub>(H) = 1.2*e*<sub>q</sub>(C) for CH or 1.5*U*<sub>eq</sub>(C,O) for CH<sub>3</sub> and OH.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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